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$(PMePh_3)(7,8-Et_2-7,8-nido-C_2B_9H_{10})$

REIJO SILLANPÄÄ, JOSEFINA PEDRAJAS, CLARA VIÑAS, FRANCES TEIXIDOR AND RAIKKO KIVEKÄS (

^aDepartment of Chemistry, University of Turku, FIN-20014 Turku, Finland, ^bInstitut de Ciéncia de Materials de Barcelona (CSIC), Campus UAB, 08193 Bellaterra. Spain. and ^cDepartment of Chemistry, University of Helsinki, FIN-00014 Helsinki, Finland. E-mail: sillanpa@utu.fi

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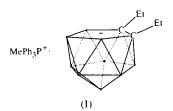
Abstract

Methyltriphenylphosphonium 7,8-diethyl-7,8-dicarbanido-undecaborate(1-), $C_{19}H_{18}P^+\cdot C_6H_{20}B_9^-$, is a salt which is formed from the cationic phosphonium ion and anionic nido carborane cage. The C—C distance in the boron cage is 1.567 (4) Å.

Comment

Our studies concerning *closo* - 1, 2 - dicarbadodecaborane (12)s have led us to conclude that the C—C distance in the cage can be modified (Sillanpää *et al.*, 1996). In the case of relevant *nido* cages, the elongation seems to be less striking. For example, in *nido*-(7,8-C₂B₉H₁₂)⁻ the C—C distance is 1.542 (3) Å (Buchanan *et al.*, 1990) and in *nido*-(7,8-Ph₂-7,8-C₂B₉H₁₀)⁻ (NHEt³/₃ salt) 1.590 (5) Å and for [NMe₃(CH₂Ph)]⁺ salt 1.602 (3) Å (Cowie *et al.*, 1993). In order to get more

information on the dependence of the C—C distance as a function of the C substituent, the crystal structure of (PMePh₃)(7,8-Et₂-7,8-C₂B₉H₁₀), (I), is now reported.



An anionic structural unit of the salt is shown in Fig. 1. The methyl group of one ethyl arm bonded to the cage C atom is orientated up from the upper belt (atoms C7, C8, B9, B10 and B11) of the boron cage. The methyl group of the other ethyl group is approximately in the same plane of the upper-belt atoms. The bond angles and distances in the anion and cation are normal, but some distortions are found in the anion.

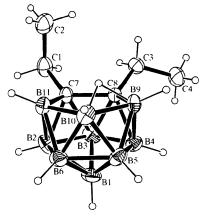


Fig. 1. ORTEPII (Johnson, 1976) plot of the anionic part of (1) in the asymmetric unit. Only the heavy atoms have been labelled. Displacement ellipsoids are drawn at the 20% probability level.

In the lower belt (B2, B3, B4, B5 and B6), the B—B distances are in the range 1.730 (7)–1.806 (7) Å (Table 1), while in the upper belt the C7—C8 bond length is 1.567 (4) (the shortest bond) and the B9—B10 bond length is 1.843 (7) Å (the longest bond). Bond angles in the lower belt range from 106.0 (3) to 109.8 (3)° and in the upper belt from 101.5 (3) to 114.5 (3)°.

Alkyl substituents on C atoms in 7,8-dicarbaboranes seem to have quite a small effect on the C—C bond distance. Aromatic substituents, compared with H substituents, lengthen the C—C distance by about 0.05 Å. Generally this follows with that which has been found for relevant *closo* cages, but in the *nido* case this effect is much weaker.

Experimental

The title compound was prepared using a method described by Viñas et al. (1997).

Crystal data

$C_{19}H_{18}P^{+}\cdot C_{6}H_{20}B_{9}^{-}$	Mo $K\alpha$ radiation
$M_r = 466.81$	$\lambda = 0.71069 \text{ Å}$
Monoclinic	Cell parameters from 20
$P2_1/c$	reflections
a = 11.6141 (15) Å	$\theta = 6.6-9.2^{\circ}$
b = 15.9238 (14) Å	$\mu = 0.114 \text{ mm}^{-1}$
c = 15.115(2) Å	T = 294 (2) K
$\beta = 100.31 (1)^{\circ}$	Prism
V = 2750.3 (6) Å ³	$0.38 \times 0.36 \times 0.26 \text{ mm}$
Z = 4	Colourless
$D_x = 1.127 \text{ Mg m}^{-3}$	

Data collection

 D_m not measured

Rigaku AFC-5S diffractom-	$R_{\rm int}=0.139$
eter	$\theta_{\text{max}} = 25.09^{\circ}$
ω –2 θ scans	$h = 0 \rightarrow 13$
Absorption correction: none	$k = 0 \rightarrow 18$
5449 measured reflections	$l = -18 \rightarrow 18$
4836 independent reflections	3 standard reflections
2427 reflections with	every 150 reflections
$I > 2\sigma(I)$	intensity decay: none

Refinement

Rejinemeni	
Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0485P)^2$
$R[F^2 > 2\sigma(F^2)] = 0.066$	+ 0.9040 <i>P</i>]
$wR(F^2) = 0.149$	where $P = (F_o^2 + 2F_c^2)/3$
S = 1.006	$(\Delta/\sigma)_{max} < 0.001$
4836 reflections	$\Delta \rho_{\text{max}} = 0.231 \text{ e Å}^{-3}$
356 parameters	$\Delta \rho_{\min} = -0.274 \text{ e Å}^{-3}$
H atoms treated by a	Extinction correction: none
mixture of independent	Scattering factors from
and constrained refinement	International Tables for
	Crystallography (Vol. C)

Table 1. Selected geometric parameters (Å, °)

P—C19	1.780 (3)	B2—B6	1.734 (7)
P—C25	1.782 (3)	B2—B3	1.744 (7)
P—C12	1.787 (3)	B3—B4	1.730 (7)
P—C13	1.795 (3)	B4—B5	1.745 (6)
C7—C8	1.567 (4)	B5—B6	1.807 (7)
C7—B11	1.610 (5)	B9—B10	1.843 (7)
C8—B9	1.620 (5)	B10—B11	1.784 (6)
C1—C7—C8 C8—C7—B11 C3—C8—C7 C7—C8—B9 B6—B2—B3 B4—B3—B2	119.1 (3) 114.5 (3) 118.3 (3) 110.6 (3) 109.1 (4) 107.7 (3)	B3—B4—B5 B4—B5—B6 B2—B6—B5 C8—B9—B10 C7—B11—B10	109.8 (3) 106.0 (3) 107.3 (3) 106.7 (3) 106.5 (3)

H atoms bonded to C were fixed and the H atoms bonded to B were refined with isotropic displacement parameters.

Data collection: MSCAFC Diffractometer Control Software (Molecular Structure Corporation, 1995). Cell refinement: MSC/AFC Diffractometer Control Software (Molecular Structure Corporation, 1995). Data reduction: TEXSAN (Molecular Structure Corporation, 1989). Program(s) used to solve structure: SHELXS86 (Sheldrick, 1985). Program(s) used

to refine structure: *SHELXL*97 (Sheldrick, 1997. Molecular graphics: *ORTEPII* (Johnson, 1976). Software used to prepare material for publication: *SHELXL*97.

Supplementary data for this paper are available from the IUCr electronic archives (Reference: OS1036). Services for accessing these data are described at the back of the journal.

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(1*R*,2*R*,3*S*,4*S*,5*S*,6*S*)-*exo*-2-Cyano-*exo*-3-[(*S*)-1,2-dibenzyloxyethyl]-*exo*-5-iodo-bicyclo[2.2.1]heptane-*endo*-2,6-carbolactone

MIRIAM ALÍAS, ELENA BUÑUEL, CARLOS CATIVIELA, MARÍA D. DÍAZ-DE-VILLEGAS AND JOSÉ A. GÁLVEZ

Departamento de Química Orgánica, Instituto de Ciencia de Materiales de Aragón, Universidad de Zaragoza-CSIC, 50009 Zaragoza, Spain. E-mail: jagl@posta.unizar.es

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Abstract

In the title compound ($C_{25}H_{24}INO_4$), an enantiomerically pure iodolactone, the absolute configurations at the chiral C1, C2, C3, C4, C5 and C6 centres have been unambiguously assigned as 1R, 2R, 3S, 4S, 5S and 6S, respectively. In the bicyclo[2.2.1]heptane (norbornane) unit, the six-membered ring presents a boat conformation, and is fused with a five-membered lactone ring which adopts an envelope conformation.